

PHYSICAL CONSTANTS

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| | $\pi = 3.142$ |
| Charge of an electron | $-e = -1.6 \times 10^{-19} \text{ C}$ |
| | $1 \text{ eV} = 1.6 \times 10^{-19} \text{ J (N}\cdot\text{m or W}\cdot\text{s)}$ |
| Permittivity of free space, ϵ_0 , gives | $1/4\pi\epsilon_0 = 9 \times 10^9 \text{ F}^{-1}\cdot\text{m}$ |
| Permeability of free space | $\mu_0 = 4\pi \times 10^{-7} \text{ H}\cdot\text{m}^{-1}$ |
| | $e^2/4\pi\epsilon_0 = 1.44 \times 10^{-9} \text{ eV}\cdot\text{m}$ |
| Speed of light in free space | $c = 3.00 \times 10^8 \text{ m}\cdot\text{s}^{-1}$ |
| | $= 1/\sqrt{\mu_0\epsilon_0}$ |
| Planck's constant | $h = 6.63 \times 10^{-34} \text{ J}\cdot\text{s}$ |
| | $h/2\pi = \hbar = 1.05 \times 10^{-34} \text{ J}\cdot\text{s}$ |
| Boltzmann's constant | $k = k_B = 1.38 \times 10^{-23} \text{ J}\cdot\text{K}^{-1}$ |
| | $= 8.63 \times 10^{-5} \text{ eV}\cdot\text{K}^{-1}$ |
| Universal Gas constant | $R = N_A \times k_B = 8.31 \text{ J}\cdot\text{K}^{-1}\text{mol}^{-1}$ |
| Molar volume at STP | $= 22.4 \times 10^{-3} \text{ m}^3$ |
| Standard Atmosphere | $= 1.01 \times 10^5 \text{ Pa}$ |
| Stefan-Boltzmann constant | $\sigma = 5.67 \times 10^{-8} \text{ W}\cdot\text{m}^{-2}\cdot\text{K}^{-4}$ |
| Wien's displacement constant | $b = 2.9 \times 10^{-3} \text{ m}\cdot\text{K}$ |
| Gravitational constant | $G = 6.67 \times 10^{-11} \text{ m}^3\cdot\text{kg}^{-1}\cdot\text{s}^{-2}$ |
| Acceleration due to gravity at sea level | $g = 9.81 \text{ m}\cdot\text{s}^{-2}$ |
| Rest mass of an electron | $m_e = 9.11 \times 10^{-31} \text{ kg}$ |
| Rest mass of a proton | $m_p = 1.67 \times 10^{-27} \text{ kg}$ |
| Avogadro's number | $N_A = 6.02 \times 10^{26} \text{ kmol}^{-1}$ |
| | $= 6.02 \times 10^{23} \text{ mol}^{-1}$ |
| Bohr magneton | $\mu_b = 9.27 \times 10^{-24} \text{ A}\cdot\text{m}^2 \text{ (or J}\cdot\text{T}^{-1})$ |
| | $= e\hbar/2m_e$ |

$$1 \text{ amu} = \frac{1}{12} M_{\text{Carbon}} = 931.5 \text{ MeV}/c^2 = 1.66 \times 10^{-27} \text{ kg}$$

$$1 \mu\text{m} = 10^{-6} \text{ m} \quad 1 \text{ nm} = 10^{-9} \text{ m} \quad 1 \text{ \AA} = 10^{-10} \text{ m} \quad 1 \text{ f} = 10^{-15} \text{ m}$$

$$\text{For } |e^a| < 1, \sum_{r=0}^{\infty} (e^a)^r = \frac{1}{1-e^a}.$$

Stirling's approximation for $N \gg 1$, is $\ln N! \simeq N \ln N - N$, or $N! \simeq \left(\frac{N}{e}\right)^N$, and thus $\frac{d \ln N!}{dx} \simeq \ln N \frac{dN}{dx}$.

For a spinless particle, the number of states with momentum between p and $p + dp$ in a macroscopic volume V is $D(p)dp$ where the density of states $D(p) = \left(\frac{\partial g}{\partial p}\right) = \frac{V}{h^3} 4\pi p^2$.

The 'Thermodynamic Identity' may be stated as $dU = TdS - pdV + \mu dN$, OR (as $F = U - TS$), $dF = -SdT - pdV + \mu dN$.